

A Review of Analytical Methods for Determination of Pravastatin in Pharmaceutical Dosage Forms and Biological Fluids

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ABSTRACT

Pravastatin is an inhibitor of HMG-CoA reductase inhibitor which is used as a hypolipidemic agent to reduce cholesterol level. Chemically, 9-Fluoro-11 β ,17-dihydroxy-16 α -methyl-3,20-dioxopregna-1,4-dien-21-yl disodium phosphate. Pravastatin is a drug of choice for the cardiovascular disease. It reduces the coronary and cerebrovascular morbidity and mortality in middle aged individual. Elevated plasma concentration of C-reactive protein are associated with protein increased cardiovascular disease, long term therapy with pravastatin an agent that reduces cardiovascular risk, might alter levels of this inflammatory parameter. This review consist of various analytical methods for the determination of pravastatin in various marketed pharmaceutical formulation of biological fluid. Analytical methods consist of various chromatographic methods, spectrophotometer methods and electrical methods reported for determination of pravastatin.

Keywords: Pravastatin, UV, HPLC, RP-HPLC

INTRODUCTION ^[1,2]

STRUCTURAL FORMULA:

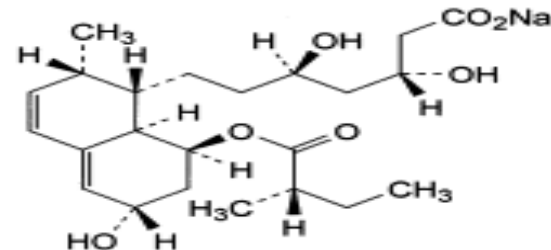


Figure No.1: structure of pravastatin

MOLECULAR FORMULA: C₂₃H₃₅NaO₇

MOLECULAR WEIGHT: 446.5gm/mol

CHEMICAL NAME: 9-Fluoro-11 β ,17-dihydroxy-16 α -methyl-3,20-dioxopregna-1,4-dien-21-yl disodium phosphate

CATEGORY: HMG Co-A reductase inhibitor, lipid regulating drug, anticholesteremic agent

DOSE: 40mg daily

DESCRIPTION: White to yellowish–white powder or crystalline powder, hygroscopic
SOLUBILITY: Freely soluble in water and in methanol, soluble in anhydrous ethanol

PHARMACOLOGICAL ACTION ^[2]:

Pravastatin is structurally similar to the HMG, a substituent of the endogenous substrate of HMG-CoA reductase. Unlike its parent compound, mevastatin, and statins such as lovastatin and simvastatin, pravastatin does not need to be activated *in vivo*. Its hydrolyzed lactone ring mimics the tetrahedral intermediate produced by the reductase allowing the agent to bind with a much greater affinity than its natural substrate. The bicyclic portion of pravastatin binds to the coenzyme A

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portion of the active site. Pravastatin sodium produces its lipid-lowering effect in two ways. First, as a consequence of its reversible inhibition of HMG-CoA reductase activity, it effects modest reductions in intracellular pools of cholesterol. This results in an increase in the number of LDL-receptors on cell surfaces and enhanced receptor-mediated catabolism and clearance of circulating LDL. Second, pravastatin inhibits LDL production by inhibiting hepatic synthesis of VLDL, the LDL precursor.

PHARMACOKINETIC ACTION ^[2]

Pravastatin is rapidly absorbed with peak plasma levels of the parent compound achieved 1 to 1.5 hours after administration. The average oral absorption of pravastatin is 34% and absolute bioavailability is 17%. These values however, are variable. Food decreases the systemic bioavailability but the lipid-lowering effect is not impacted. It is metabolised by liver and its excretion by urine (20%) and feces (70%).

SIDE EFFECT ^[3]

Difficulty with moving, muscle or bone pain, muscle stiffness, chest pain or discomfort, chills, cough, dark-colored urine, diarrhea, difficult or labored breathing, fever, headache, loss of appetite, muscle cramps or spasms, muscular tenderness, wasting, or weakness, nasal congestion, nausea, runny nose, shivering, sneezing, sore throat, sweating, swollen joints, tightness in the chest, trouble with sleeping, unusual tiredness or weakness, vomiting

ANALYTICAL METHODS

This all are the methods which are used for the determination of Pravastatin in marketed formulation and in biological fluids. This all analytical methods are reported which are seen during the literature survey. This article describes the review on the all reported analytical methods with specific conditions.

I. COMPENDIAL METHODS:

Table No.1: Summary of official compendial methods

Drug	Compendia	Column	Mobile phase	Flow rate	Wavelength
Pravastatin	Indian pharmacopoeia 2010 ^[4]	15cmx4.6m m	glacial acetic acid: triethylamine: methanol: water (1:1:450:550 v/v/v/v)	1.3ml/min	238nm
	European pharmacopoeia ^[5]	0.15cmx4.6 mm	glacial acetic acid: triethylamine: methanol: water(1:1:450:550 v/v/v/v)	1.3ml/min	238nm

II. UV SPECTROSCOPIC METHOD:

A simple, precise and economical spectrophotometric method for the estimation of Pravastatin in pharmaceutical bulk and tablet dosage form was developed and validated. Identification was carried out using a UV-visible double beam spectrophotometer detector with working wavelength that 240nm in water and

methanol medium. The method was validated with respect to its specificity, linearity range, accuracy, and precision in analytical media. Pravastatin show the maximum absorbance (λ_{max}) at 240nm. Simple UV spectroscopy, first derivative spectroscopy, AUC method and absorption ratio methods are reported for determination of the Pravastatin in marketed

formulation. Below in table describes the method description and condition which are various chromatographic methods with the reported on review literature.

Table No.2: Summary of UV spectroscopic methods of Pravastatin

Title	Method	Solvent	λ_{max} (nm)	R^2
Development of spectrophotometric method for pravastatin sodium in bulk and tablet formulation ^[6]	Simpeuv-spectroscopy	Distilled water	240nm	0.9999
Development of new analytical methods and their validation for the determination of pravastatin sodium in bulk and marketed formulation ^[7]	Indirect method	Distilled water	440nm	0.9986
	Oxidation coupling	Distilled water	627nm	0.9991
Simultaneous UV-spectrophotometric estimation of pravastatin and co-enzyme Q10 in their formulated combined dosage form and synthetic mixture ^[8]	First order derivative	Methanol:IPA(50:50V/V)(PRAVA)	236nm	0.999
		Isopropyl:alcohol(50:50V/V)(Q10)	288nm	0.999

III. CHROMATOGRAPHIC METHODS:

Various chromatographic methods are used for the determination of the Pravastatin alone or combination with other drugs in various marketed formulation and in biological fluids like human plasma and urine. Chromatographic methods like High performance liquid chromatography (HPLC/RP-HPLC), High performance thin layer chromatography (HPTLC) with UV detection are used. In which the stationary phase commonly used is C18 column and commonly used wavelength for detection is 240nm. Mobile phase is varies with condition of method in various proportion. Below in table describes the summary of the various chromatographic methods are used with the method description.

Table No.3: Summary of Chromatographic Methods of Pravastatin

Title	Method	Mobile phase	Stationary phase	Wavelength
Development, optimization and validation of HPLC method for determination of pravastatin sodium in tablet ^[9]	HPLC	methanol : water : triethylamine : glacial acetic acid in the ratio of	Reverse phase C ₁₈	238nm

		455: 545: 2: 1.2 v/v/v/v		
Reverse phase HPLC method for determination of pravastatin in tablet dosage forms ^[10]	RP-HPLC	Acetonitrile : potassium dihydrogen orthophosphate(0.02M)(30:70v/v)	Phenomenexlu na 5µm C ₁₈ (150x4.6mm)	240nm
High-performance liquid chromatography determination of pravastatin in plasma ^[11]	HPLC	Methanol: water(50:30)	Isocratic c ₁₈	240nm
Development and validation of analytical method for pravastatin ^[12]	HPLC	acetonitrile:methanol:0.08M orthophosphoric acid (23:20:57v/v/v)	hypersil ODS,3µm,10cmx4.6mm	234nm
Development of a selective LC method for the determination of pravastatin sodium ^[13]	LC	Methanol:Phosphate buffer (pH 7; 0.02 M) (57:43, v/v)	C ₁₈ (150 mm × 4.6 mm)	238nm
Method development and validation of pravastatin sodium in human plasma by using LCMS/MS ^[14]	LC/MS/MS	(80:20, v/v), acetonitrile and 2 mm ammonium formate	Hupurity advance C8, 50 x 4.6 mm, 5µm	236nm
Quantitative determination of pravastatin in pharmaceutical dosage form by HPLC with UV detection ^[15]	HPLC with UV detection	10mM ammonium acetate: methanol: triethylamine (40:60:0.17 v/v/v)	Teknokroma C ₈ (5µm,25cmx 4.6mm)	239nm
Stability study of pravastatin under hydrolytic condition assessed by HPLC ^[16]	Stability by HPLC	acetonitrile-30 mmol L ⁻¹ : phosphate buffer solution pH 2 (28:72V/V)	C ₁₈ column	239nm
Determination of pravastatin sodium and its isomeric metabolite in human urine by HPLC with UV detection ^[17]	HPLC with UV detection	Acetonitrile: methanol: water(40:50:10)	C ₁₈	240nm
Stability indicating RP-HPLC with UV detection for estimation of ezetimibe and pravastatin ^[18]	HPLC with UV detection	Acetonitrile: 0.1 % Formic acid: Methanol (40:50:10v/v/v)	C ₁₈	248nm
RP-HPLC determination of three anti-hyperlipidemic drug in spiked human plasma and in	RP-HPLC	Acetonitrile:50 mM KH ₂ PO ₄ (40:60 v/v)	150 mmx4.6 mm Zorbax Extend-C ₁₈	230nm

dosage form ^[19]			column	
Development and validation of a simple and fast HPLC method for determination of lovastatin, pravastatin and simvastatin ^[20]	HPLC	Acetonitrile and 0.1% phosphoric acid (65:35v/v)	C ₈ encapped(250x4mmx5µm), isocratic	238nm
An ultra sensitive and selective LC-UV method for the simultaneous determination of pravastatin, diltiazem. Naproxen sodium and meloxicam in API pharmaceutical formulation and human serum ^[21]	LC-UV	Methanol:water (80:20 v/v)	Purospher Star, C ₁₈ (5 µm, 25 x 0.46 cm)	220nm
Determination of simvastatin, pravastatin sodium and rosuvastatin calcium in tablet dosage form by HPTLC ^[22]	HPTLC	Chloroform :methanol :toluene (6:2:2, v/v/v)	Precoated silica gel 60F 254	236nm
The application of microbore UPLC/oa-TOF-MS and 1H NMR spectroscopy to the metabonomic analysis of rat urine following the intravenous administration of pravastatin ^[23]	UPLC/oa-TOF-MS and 1H NMR	Methanol: acetonitrile: water(50:40:30v/v)	1mm i.d. x 100 mm column	240nm

IV. MISCELLANEOUS METHODS:

Most widely used methods are mainly HPLC, UV and HPTLC for determination of pravastatin in various formulation or in biological fluids but along with that other methods are also used which are seen during the literature survey. The summary of that methods are described below in table.

Table No.4: Summary of Miscellaneous methods of Pravastatin

Sr no	Title	Method
1	Identification of an impurity in pravastatin by application of collision-activated decomposition mass spectra ^[24]	Identification of impurity
2	Capillary electrophoresis determination of pravastatin and separation of its degradation product ^[25]	Capillary electrophoresis

3	Kinetic spectrophotometric determination of pravastatin in drug formulation via derivatization with 4-chloro-7-nitrobenzo-2-oxa-1,3-diazole(NBD-Cl) ^[26]	Kinetic spectroscopy
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CONCLUSION

The presented review highlights on various analytical methods reported on Pravastatin and in combination with other drug. HPLC-HPTLC-UV methods were found to be most widely used. Various chromatographic conditions are presented in under Table. The faster time, high sensitivity; specificity and better separation efficiency enable HPLC to be used frequently for the determination of Pravastatin in the comparison with the other methods. There is

no doubt on the fact that these chromatographic methods are rapid and far more economical. Other methods are also useful. In this way various analytical methods for the estimation of Pravastatin in bulk or in various matrixes like plasma, alone or in combination with other drugs is discussed. The presented information is useful for the researchers especially those involved in the formulation development and quality control of Pravastatin in combination with other drug.

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